## **Alkane Properties.**

Name	Formula	B.p., °C	Density, g/mL
Methane	CH <sub>4</sub>	-162	gas
Ethane	CH <sub>3</sub> CH <sub>3</sub>	-90	gas
Propane	CH <sub>3</sub> CH <sub>2</sub> CH <sub>3</sub>	-42	gas
n-Butane	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>2</sub> CH <sub>3</sub>	0	gas
n-Pentane	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>3</sub> CH <sub>3</sub>	36	0.626
n-Hexane	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>4</sub> CH <sub>3</sub>	69	0.659
n-Heptane	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>5</sub> CH <sub>3</sub>	98	0.684
n-Octane	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>6</sub> CH <sub>3</sub>	126	0.703
Isomers: C <sub>4</sub> H <sub>10</sub>			
n-Butane	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>2</sub> CH <sub>3</sub>	0	gas
2-Methyl propane	(CH <sub>3</sub> ) <sub>2</sub> CHCH <sub>3</sub>	-12	gas
Isomers: C <sub>5</sub> H <sub>12</sub>			
n-Pentane	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>3</sub> CH <sub>3</sub>	36	0.626
2-Methyl butane	(CH <sub>3</sub> ) <sub>2</sub> CHCH <sub>2</sub> CH <sub>3</sub>	28	0.620
2,2-Dimethyl propane	(CH <sub>3</sub> ) <sub>4</sub> C	10	0.615
Isomers: C <sub>6</sub> H <sub>14</sub>			
n-Hexane	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>4</sub> CH <sub>3</sub>	69	0.659
2-Methyl pentane	(CH <sub>3</sub> ) <sub>2</sub> CH(CH <sub>2</sub> ) <sub>2</sub> CH <sub>3</sub>	60	0.654
3-Methyl pentane	CH <sub>3</sub> CH <sub>2</sub> CH(CH <sub>3</sub> )CH <sub>2</sub> CH <sub>3</sub>	63	0.676
2,2-Dimethyl butane	(CH <sub>3</sub> ) <sub>3</sub> CCH <sub>2</sub> CH <sub>3</sub>	50	0.649
2,3-Dimethyl butane	(CH <sub>3</sub> ) <sub>2</sub> CHCH(CH <sub>3</sub> ) <sub>2</sub>	58	0.668